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**CO-ORDINATED SCIENCES**

**0654/53**

Paper 5 Practical Test

**October/November 2017**

MARK SCHEME

Maximum Mark: 45

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**Published**

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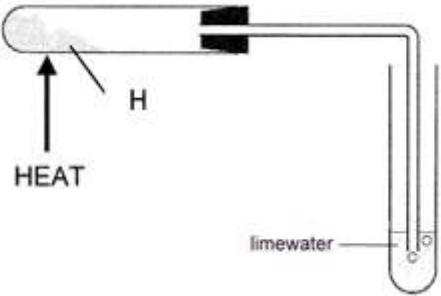
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Question	Answer	Marks
1(a)(i)	four readings ; to the nearest 0.1 cm ; decreasing ;	<b>3</b>
1(a)(ii)	cm <sup>2</sup> and cm ;	<b>1</b>
1(b)(i)	axes correctly labelled with units ; suitable linear scale using at least half the grid ; all 4 points plotted correctly $\pm$ half small square ; best-fit straight line through origin ;	<b>4</b>
1(b)(ii)	more enzyme faster reaction ORA ;	<b>1</b>
1(b)(iii)	correct reading from graph ; lines on graph to show working ;	<b>2</b>
1(c)	glowing splint <b>and</b> relights;	<b>1</b>
1(d)	any <b>two</b> from: temperature ; pH ; pieces from same potato ; concentration of peroxide ;	<b>2</b>
1(e)	only cells on the outside of the potato are in contact in the peroxide ;	<b>1</b>

Question	Answer	Marks						
2(a)(i)	 <p>either <b>H</b> or limewater label required ;</p>	1						
2(a)(ii)	<p><b>(H)</b> turns yellow (solid) ; white when solid cools ; (limewater) milky / white ppt. ; goes colourless / ppt. dissolves / ppt. disappears / less ppt. ;</p>	4						
2(b)(i)	bubbles / effervescence / colourless solution ;	1						
2(b)(ii)	white ppt. ; ppt. dissolves / ppt. disappears / forms colourless solution (from ppt.) ;	2						
2(c)	<b>(H is)</b> zinc ; carbonate ;	2						
2(d)(i)	<p>grid with all entries (as below) ;</p> <table border="1" data-bbox="349 1110 1167 1262"> <thead> <tr> <th data-bbox="349 1110 757 1161">test</th> <th data-bbox="757 1110 1167 1161">observation</th> </tr> </thead> <tbody> <tr> <td data-bbox="349 1161 757 1211">barium nitrate solution</td> <td data-bbox="757 1161 1167 1211"></td> </tr> <tr> <td data-bbox="349 1211 757 1262">silver nitrate solution</td> <td data-bbox="757 1211 1167 1262"></td> </tr> </tbody> </table>	test	observation	barium nitrate solution		silver nitrate solution		1
test	observation							
barium nitrate solution								
silver nitrate solution								

Question	Answer	Marks	
2(d)(ii)	test	observation	2
	barium nitrate solution	no reaction ;	
	silver nitrate solution	yellow ppt. ; (accept cream ppt.)	
2(d)(iii)	brown solution ;	1	
2(d)(iv)	not a sulfate / not a chloride / iodide / not a bromide ;	1	

Question	Answer	Marks
3(a)(i)	v recorded for u = 30 cm ; to the nearest 0.1 cm ;	2
3(a)(ii)	all v values present ; v values decreasing ;	2
3(a)(iii)	ratios correct ;	1
3(a)(iv)	all values correct ;	1
3(b)(i)	suitable choice of scales ( $\geq$ half the grid used) ; at least 4 plots correct to half a small square (penalise 'blobs') ; good best-fit line judgement ;	3
3(b)(ii)	indication <b>on graph</b> of how data obtained <b>and</b> at least half of line used ; correct calculation for triangle method using data from graph ;	2

Question	Answer	Marks
3(b)(iii)	(15.0 ± 1.0) cm ; 2 / 3 significant figures ;	<b>2</b>
3(c)	any <b>two</b> from: move screen slowly / to and fro (until sharpest focus obtained) ; repeat each reading <b>and</b> average ; object / lens / screen perpendicular to bench ; object and lens same height above the bench ; carry out experiment away from other bright light sources / in a darkened room / use bright(er) light source ;	<b>2</b>